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## Structure Reports

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# Xiao-Yang Qiu, ${ }^{\text {a,b }}$ Sen-Lin Yang, ${ }^{\text {a }}$ Wei-Sheng Liu ${ }^{\text {b }}$ and Hai-Liang Zhu ${ }^{\text {c }}$ * 

${ }^{\text {a }}$ Department of Chemistry, Fuyang Normal College, Fuyang Anhui 236041, People's Republic of China, ${ }^{\text {b }}$ Department of Chemistry, Lanzhou University, Lanzhou 730000, People's Republic of China, and ${ }^{\text {c }}$ Institute of Functional Biomolecules, State Key Laboratory of Pharmaceutical Biotechnology, Nanjing University, Nanjing 210093, People's Republic of China

Correspondence e-mail: liuws@lzu.edu.cn, hailiang_zhu@163.com

## Key indicators

Single-crystal X-ray study
$T=298 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.053$
$w R$ factor $=0.139$
Data-to-parameter ratio $=17.2$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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## (E)-3-(4-Hydroxyphenyl)-1-(4-methoxy-phenyl)prop-2-en-1-one

In the title compound, $\mathrm{C}_{16} \mathrm{H}_{14} \mathrm{O}_{3}$, the dihedral angle between the two benzene rings is $25.8(2)^{\circ}$. Molecules are linked into ribbons through $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}=\mathrm{C}$ hydrogen bonds.

## Comment

Recently, we have reported the structures of a few chalcone derivatives (Qiu et al., 2006, 2006a), including the analogous chloro derivative (Qiu et al., 2006b). As an extension of our work on the structural characterization of chalcone derivatives, the title compound, (I), is reported here.

(I)

In (I), all bond lengths are within normal ranges (Allen et al., 1987) (Fig. 1). The $\mathrm{C} 8=\mathrm{C} 9$ bond length of 1.328 (3) $\AA$ conforms to the value for a $\mathrm{C}=\mathrm{C}$ double bond. The dihedral angle between the two benzene rings is 25.8 (2) ${ }^{\circ}$. In the crystal structure, molecules are linked through intermolecular O $\mathrm{H} \cdots \mathrm{O}=\mathrm{C}$ hydrogen bonds, forming ribbons along the $a$-axis direction (Table 1 and Fig. 2).

## Experimental

The reagents were commercial products and were used without further purification. An aqueous solution of potassium hydroxide $(5 \%, 1 \mathrm{ml})$ was added with stirring overnight to a solution of 4 hydroxybenzaldehyde ( $1 \mathrm{mmol}, 0.12 \mathrm{~g}$ ) and 4-methoxyacetophenone $(1 \mathrm{mmol}, 0.15 \mathrm{~g})$ in ethanol $(15 \mathrm{ml})$ at room temperature. The reaction mixture was then poured on to ice and neutralized with aqueous hydrochloric acid ( $5 \%$ ). A white solid was prepared after neutralization and obtained from an ethanol solution. The solid $(0.05 \mathrm{mmol}$,


Figure 1
The structure of (I), showing displacement ellipsoids at the $30 \%$ probability level for non-H atoms and the atom-numbering scheme.

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0.013 g ) was dissolved in acetone ( 12 ml ) and stirred for about 10 min to give a clear colourless solution. After keeping the solution in air for 9 d , colourless plate-shaped crystals were formed at the bottom of the vessel on slow evaporation of the solvent. These were collected, washed three times with acetone and dried in a vacuum desiccator using $\mathrm{CaCl}_{2}$ (yield $53 \%$ ).

## Crystal data

$$
\begin{array}{ll}
\mathrm{C}_{16} \mathrm{H}_{14} \mathrm{O}_{3} & Z=8 \\
M_{r}=254.27 & D_{x}=1.296 \mathrm{Mg} \mathrm{~m}^{-3} \\
\text { Orthorhombic, } P b c a & \text { Mo } K \alpha \text { radiation } \\
a=13.3482(9) \AA & \mu=0.09 \mathrm{~mm}^{-1} \\
b=7.6922(5) \AA & T=298(2) \mathrm{K} \\
c=25.3822(17) \AA & \text { Plate, colourless } \\
V=2606.2(3) \AA^{3} & 0.34 \times 0.17 \times 0.07 \mathrm{~mm}
\end{array}
$$

## Data collection

Bruker SMART APEX CCD diffractometer
$\omega$ scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.978, T_{\text {max }}=0.986$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.053$
$w R\left(F^{2}\right)=0.139$
$S=1.00$
2978 reflections
173 parameters
H -atom parameters constrained

$$
\begin{aligned}
& w=1 /[ \sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.057 P)^{2} \\
&+0.0775 P] \\
& \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.14 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.17 \mathrm{e}^{-3}
\end{aligned}
$$

14929 measured reflections 2978 independent reflections 1485 reflections with $I>2 \sigma(I)$ $R_{\text {int }}=0.067$ $\theta_{\text {max }}=27.6^{\circ}$

Table 1
Hydrogen-bond geometry $\left(\AA^{\circ},^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| O3-H13 $^{\cdots} \mathrm{O}^{\mathrm{i}}$ | 0.82 | 1.89 | $2.651(2)$ | 154 |

Symmetry code: (i) $x+\frac{1}{2}, y,-z+\frac{3}{2}$.
H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with $\mathrm{C}-\mathrm{H}$ and $\mathrm{O}-\mathrm{H}$ distances of $0.93-0.96$ and $0.82 \AA$, respectively, and with $U_{\text {iso }}(\mathrm{H})=$ $1.2 U_{\mathrm{eq}}(\mathrm{C})$ or $1.5 U_{\mathrm{eq}}(\mathrm{C}, \mathrm{O})$.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve


Figure 2
The crystal packing viewed along the $b$ axis, showing the intermolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{O}$ hydrogen bonds (dashed lines) linking the molecules into ribbons running along $a$.
structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

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## References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. \& Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-19.

Bruker. (1998). SMART (Version 5.628) and SAINT (Version 6.02), Bruker AXS Inc., Madison, Wisconsin, USA.
Qiu, X.-Y., Liu, W.-S. \& Zhu, H.-L. (2006). Acta Cryst. E62, o1304-o1305.
Qiu, X.-Y., Yang, S.-L., Liu, W.-S. \& Zhu, H. (2006a). Acta Cryst. E62, o1627o1628.
Qiu, X.-Y., Yang, S.-L., Liu, W.-S. \& Zhu, H.-L. (2006b). Acta Cryst. E62, o2685-o2686.
Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Sheldrick, G. M. (1997b). SHELXTL. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.


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